

POWER REQUIREMENT FORMULAS

Formula A :

$$\text{Wattage required for heat-up} = \frac{\text{Weight of material (lbs)} \times \text{Specific Heat} (*) \times \text{Temperature Rise (}^\circ\text{F)}}{3.412 \times \text{Heat-up Time (Hours)}}$$

*For specific heat and density values, please refer to pages ((X-X))

Formula B :

$$\text{Wattage losses at operating temperature} = \text{Wattage loss/ft}^2 (**)\text{ x Area (ft}^2\text{)}$$

**To find values for wattage losses, refer to the curves shown on page ((X))

Formula C :

$$\text{Wattage for melting or vaporizing} = \frac{\text{Weight of material (lbs)} \times \text{Heat of fusion or vaporization (BTU/lb) (***)}}{3.412 \times \text{Heat-up Time (Hours)}}$$

*** The Specific Heat of a material changes at some temperature due to melting (fusion) or evaporation (vaporation). To calculate wattage, use formula A for the heat needed to rise from initial to the point of change. Then calculate using formula A with the new specific heat from the point of change to the final temperature. Add these two values to formula B and C to calculate the total wattage

Specific Calculations

To Heat Liquids :

$$\text{Wattage for initial heat up} = (A) + (B)/2$$

$$\text{Wattage for heating added material} = (A \text{ for new material}) + (B)$$

To Melt Metals :

$$\text{Wattage for initial heat up} = (A \text{ to melting point}) + (C \text{ to melt}) + (A \text{ to heat above melting point}) + (B)/2$$

$$\text{Wattage for heating added material} = (A \text{ to melting point}) + (C \text{ to melt}) + (A \text{ to heat above melting point})$$

For the two calculations above, add the two values and multiply the final wattage by 1.2 to compensate for additional heat losses (Safety Factor)

To Heat Ovens :

$$\text{Wattage} = [(A \text{ for air}) + (A \text{ for material introduced into oven}) + (B \text{ for external oven losses})] * 1.25 \text{ (door heat loss)}$$

Forced Air Heating :

$$\text{Wattage} = \frac{\text{Ft}^3/\text{min.} \times \text{Temperature Rise (}^\circ\text{F)}}{3}$$

POWER FORMULA EXAMPLE 1

Basic Heating Calculations

A mold is being heated to form rubber parts. Every hour, 100 ounces of rubber is placed into the 10" x 10" x 4" steel mold which is attached between two stainless steel platens measuring 15" x 15" x 2". Each platen is insulated on it's horizontal surface with 1/2" thick insulation.

The mold must reach an operating temperature of 400°F in one hour from room temperature (70°F).

We first find the specific heat and weight values for each of our materials:

Specific Heat of Steel:	0.12 $\frac{\text{BTU}}{\text{Lb} \times ^\circ\text{F}}$	Weight of Steel Mold: $10 \times 10 \times 4 \times .284 \frac{\text{Lb.}}{\text{In}^3} = 113.6 \times 2\text{pc} = 227.2 \text{ lbs}$
Specific Heat of S. Steel:	0.12 $\frac{\text{BTU}}{\text{Lb} \times ^\circ\text{F}}$	Weight of SS316 Platen: $15 \times 15 \times 2 \times .288 \frac{\text{Lb.}}{\text{In}^3} = 129.6 \times 2\text{pc} = 259.2 \text{ lbs}$
Specific Heat of Rubber:	0.44 $\frac{\text{BTU}}{\text{Lb} \times ^\circ\text{F}}$	Weight of Rubber: $100 \text{ ounces} \times 0.0625 = 6.25 \text{ lbs.}$

Formula A : Wattage required for heat-up

To Heat Mold:	$\frac{227.2 \text{ (lbs)} \times 0.12 \text{ (BTU/Lb}^\circ\text{F)} \times 330 \text{ (}^\circ\text{F)}}{3.412 \times 1 \text{ (hr.)}}$	=	2637 Watts
To Heat Platens:	$\frac{259.2 \text{ (lbs)} \times 0.12 \text{ (BTU/Lb}^\circ\text{F)} \times 330 \text{ (}^\circ\text{F)}}{3.412 \times 1 \text{ (hr.)}}$	=	3008 Watts
To Heat Rubber:	$\frac{6.25 \text{ (lbs)} \times 0.44 \text{ (BTU/Lb}^\circ\text{F)} \times 330 \text{ (}^\circ\text{F)}}{3.412 \times 1 \text{ (hr.)}}$	=	266 Watts
Safety Factor:	$(2637 + 3008 + 266) \times 20\%$	=	1182 Watts
Total wattage required for heat-up:			<u>7093 Watts</u>

Formula B : Wattage losses at operating temperatures:

Heat Loss from Mold (vertical surfaces):	$\frac{10'' \times 4'' \times 4 + 10'' \times 4'' \times 4}{144 \text{ in}^2/\text{ft}^2}$	=	2.2 ft ² x 350 $\frac{\text{Watts}}{\text{ft}^2}$	=	778 Watts
Heat Loss from Platen (vertical surfaces):	$\frac{15'' \times 2'' \times 4 + 15'' \times 2'' \times 4}{144 \text{ in}^2/\text{ft}^2}$	=	1.7 ft ² x 350 $\frac{\text{Watts}}{\text{ft}^2}$	=	583 Watts
Heat Loss from Platen (bare horizontal):	$\frac{(15'' \times 15'' \times 2) - (10'' \times 10'' \times 2)}{144 \text{ in}^2/\text{ft}^2}$	=	1.7 ft ² x 250 $\frac{\text{Watts}}{\text{ft}^2}$	=	434 Watts
Heat Loss from Platen (insul. horizontal):	$\frac{15'' \times 15'' \times 2}{144 \text{ in}^2/\text{ft}^2}$	=	3.13 ft ² x 100 $\frac{\text{Watts}}{\text{ft}^2}$	=	313 Watts
Safety Factor:	$(778 + 583 + 434 + 313) \times 20\%$	=			<u>422 Watts</u>
Total wattage losses at operating temperature:					2530 Watts
Total wattage required for heat-up:					7093 Watts
Total wattage required:					<u>9623 Watts</u>

POWER FORMULA EXAMPLE 2

Melting Calculation

An open top uninsulated steel tank measures 20" x 22" x 18" and weighs 140 lbs. The tank is used to melt 175 pounds of paraffin wax from room temperature (72°F) to 150°F in 3 hours.

We first find the properties of the materials we will be using:

Specific Heat of Steel:	0.12 $\frac{\text{BTU}}{\text{Lb } ^\circ\text{F}}$	Heat of Fusion of Paraffin:	63 $\frac{\text{BTU}}{\text{Lb}}$
Specific Heat of Solid Paraffin:	0.70 $\frac{\text{BTU}}{\text{Lb } ^\circ\text{F}}$	Wax Surface Loss @150°F:	55 $\frac{\text{Watts}}{\text{ft}^2}$
Specific Heat of Melted Paraffin:	0.71 $\frac{\text{BTU}}{\text{Lb } ^\circ\text{F}}$	Steel Surface Loss @150°F:	70 $\frac{\text{Watts}}{\text{Ft}^2}$
Melting Point of Paraffin:	133 °F	Wax Surface Area: (20" x 22")/144 = 3 ft ²	Tank Surface Area: [(20" + 22") x 18" x 2]/144 = 11 ft ²

Formula A : Wattage required for heat-up

To Heat Tank:	$\frac{140 \text{ (lbs)} \times 0.12 \text{ (BTU/Lb}^\circ\text{F)} \times (150-72)(^\circ\text{F)}}{3.412 \times 3 \text{ (hrs)}}$	=	128 Watts
To Heat Wax:	$\frac{175 \text{ (lbs)} \times 0.70 \text{ (BTU/Lb}^\circ\text{F)} \times (133-72)(^\circ\text{F)}}{3.412 \times 3 \text{ (hrs)}}$	=	730 Watts
To Heat Melted Wax:	$\frac{175 \text{ (lbs)} \times 0.71 \text{ (BTU/Lb}^\circ\text{F)} \times (150-133)(^\circ\text{F)}}{3.412 \times 3 \text{ (hrs)}}$	=	206 Watts
Safety Factor:	(128 + 730 + 206) x 20%	=	213 Watts
Total wattage required for heat-up:			<u>1277 Watts</u>

Formula C : Wattage required for melting

Heat of Fusion for Paraffin:	$\frac{175 \text{ (lbs)} \times 63 \text{ (BTU/lb)}}{3.412 \times 3 \text{ (hrs)}}$	=	1077 Watts
Safety Factor:	1077 x 20%	=	215 Watts
Total wattage required for melting:			<u>1292 Watts</u>

Formula B : Wattage losses at operating temperature

Heat loss from Paraffin:	3 ft ² x $\frac{70 \text{ Watts}}{\text{Ft}^2}$	=	210 Watts
Heat Loss from tank (vertical surfaces):	11 ft ² x $\frac{55 \text{ Watts}}{\text{Ft}^2}$	=	605 Watts
Safety Factor:	(210 + 605) x 20%	=	163 Watts
Total wattage losses:			978 Watts
Total wattage required for heat-up:			1277 Watts
Total wattage required for melting:			1292 Watts
Total wattage required:			<u>3547 Watts</u>